





Research Article

# Nutritional Potential, Phytochemical Study, and Antioxidant Activity of the Pulp and Seeds of Rambutan (*Nephelium lappaceum* L.) Fruits

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## Abstract

Rambutan is increasingly popular in Europe but remains underexploited in Cameroon, both locally and within the agro-food industry. The aim of this study was to evaluate the nutritional potential, phytochemical composition, and antioxidant activity of its pulp and seeds. Ripe fruits were harvested, sorted, washed, and separated into pulp and seeds. After oven-drying and grinding, the resulting powders were used for analyses. Protein, lipid, carbohydrate, and crude fiber contents were determined using standard methods. Minerals were quantified using fluorescence spectrometry. Polyphenols, flavonoids, and tannins were assessed using colorimetric methods. Antioxidant activity was evaluated *in vitro* using DPPH, ABTS, and FRAP assays, and *ex vivo* through the quantification of malondialdehyde levels and the activities of catalase and superoxide dismutase. The results showed that the pulp was characterized by a high moisture content, whereas the seed exhibited significantly higher levels of total carbohydrates (22.95g/100g FM), proteins (2.66g/100g FM), lipids (5.86g/100g FM), dietary fiber (4.59g/100g FM), and ash (10.36g/100g FM). From a mineral perspective, the seed contained higher concentrations of potassium (37mg/100g FM), calcium (10.54mg/100 g FM), iron (0.61mg/100g FM), manganese (0.59mg/100g FM), zinc (0.061mg/100g FM), and copper (0.48mg/100g FM), whereas the pulp exhibited a higher phosphorus content. Phytochemical analysis demonstrated higher concentrations of total polyphenols, flavonoids, and condensed tannins in the seeds than in the pulp. The *in vitro* antioxidant activity confirmed the reducing power and free radical scavenging ability of the extracts, as demonstrated by the FRAP, DPPH, and ABTS assays. *Ex vivo* assays revealed an increase in antioxidant enzyme activity alongside a reduction in malondialdehyde levels, suggesting a protective effect against oxidative stress. Rambutan is rich in nutritional and bioactive compound, making it a promising resource for dietary supplements.

## Keywords

Nutritional Potential, Phytochemical Compound, Antioxidant Activity, *Rambutan*

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Received: 29 May 2026; Accepted: 12 June 2026; Published: 30 June 2026



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## 1. Introduction

Fruits play an essential role in a balanced diet [1] and are associated with a reduced risk of premature mortality and overall improved health [2]. Among these fruits is rambutan (*Nephelium lappaceum L.*), a tropical fruit belonging to the Sapindaceae family. *Nephelium lappaceum L.* has an ovoid to spherical shape, with a leathery skin covered in flexible spines, and its color ranges from red to yellow, and even green [3, 4].

From a nutritional standpoint, the fruits of *Nephelium lappaceum L.* contain high amounts of carbohydrates, proteins, and lipids, as well as several minerals such as potassium, calcium, magnesium, and iron. This nutritional richness gives them an interesting energy and functional value for human consumption [5]. They are also rich in dietary fiber, vitamin C, and phenolic compounds, and are distinguished by their antioxidant properties [6, 7]. *Nephelium lappaceum L.* also contains various secondary metabolites, including polyphenols, flavonoids, tannins, and saponins, which are recognized for their protective effects against oxidative stress [3]. Among these, gallic acid, geraniin, and corilagin are among the main compounds responsible for the fruit's antioxidant activity. Their levels vary depending on the part of the fruit [8]. Thus, the fruits of *Nephelium lappaceum L.* have unique nutritional compositions and are rich sources of bioactive compound, contributing to beneficial health effects [9].

Traditionally, *Nephelium lappaceum L.* is widely used in traditional medicine, particularly in Southeast Asia, for the treatment of various disorders. The entire plant bark, leaves, roots, and seeds is utilized for its medicinal properties. In Malaysia, dried bark is traditionally used to manage diabetes and hypertension [10]. Root decoctions are used to reduce fever, while the seeds are sometimes employed as a remedy for diarrhea. These traditional uses also include the treatment of gastrointestinal disorders, skin diseases, dysentery, and giardiasis [8, 11, 12]. In addition to its traditional uses, the fruit of *Nephelium lappaceum L.* exhibits important therapeutic properties and is recognized for its antidiarrheal, antidiabetic, anthelmintic, astringent, and carminative effects [13]. Furthermore, studies have reported anti-arthritis and anti-inflammatory effects of the ethanolic bark extract in rats [9], as well as anti-cancer properties associated with the inhibition of certain tumor cell lines [14].

Despite these benefits, *Nephelium lappaceum L.* remains underutilized in Cameroon in terms of consumption, processing, and domestication, and there is a lack of comprehensive research on this fruit. The aim of this study was to determine the nutritional composition, phytochemical compound, and antioxidant activity of rambutan (*Nephelium lappaceum L.*) pulp and seeds, with the goal of valorizing this fruit by providing nutritional information to promote its consumption within dietary and therapeutic contexts.

## 2. Materials and Methods

### 2.1. Plant Material

The rambutan fruit was identified at the National Herbarium of Cameroon under No. 67584/HNC as *Nephelium lappaceum L.*, by comparison with the original material consisting of branches, leaves, stem, and fruit.

### 2.2. Preparation of Plant Material

Ripe and fresh fruits of *Nephelium lappaceum L.* were harvested from fields in the town of Njombe. Upon arrival at the laboratory, the fruits were sorted and manually cleaned using a sterile kitchen knife. Pulp and seed samples were oven-dried at 50°C for 8 hours and 12 hours, respectively. The dried samples were then ground using a BLINDER brand food processor. The ground samples were packaged in sealed kraft paper bags for analysis.

### 2.3. Determination of Moisture Content

Moisture content was determined according to the AOAC method [12], which is based on measuring the mass of water lost from pulp and seed samples after oven-drying at 105°C until a constant dry weight was achieved.

### 2.4. Determination of Macronutrient Content

The contents of proteins, lipids, total carbohydrates, and dietary fibers were determined in the pulp and seeds of *Nephelium lappaceum*.

#### 2.4.1. Protein Content

Total nitrogen was determined using the Kjeldahl method [16]. This method involved three stages, namely digestion, distillation, and titration. These steps allowed the determination of total nitrogen content in the different samples. Protein content was then calculated by multiplying the nitrogen content by the conversion factor 6.25.

#### 2.4.2. Lipid Content

Lipid content was determined using the Soxhlet extraction method with hexane, following the AOAC procedure [15]. A known mass of pulp and seed powder was placed in an extraction thimble and then inserted into a Soxhlet extractor containing hexane. After complete extraction and cooling, the cellulose thimble was removed from the extractor and the residual solvent was transferred into a flask. A rotary evaporator (Büchi Rotavapor R-205) was used to concentrate the oil contained in the samples, and lipid content relative to dry mass was determined gravimetrically.

### 2.4.3. Crude Fiber Content

Dietary fiber content was determined using the enzymo-gravimetric method [16], based on digestion and filtration steps. A sample mass was treated with an  $\alpha$ -amylase solution at 90°C under agitation for 15 minutes. Proteins were then digested using protease, and residual starch was degraded with amyloglucosidase. After precipitation, the residue was washed with ethanol and acetone, followed by filtration using filter paper. The crude fiber content was then determined gravimetrically.

### 2.4.4. Carbohydrate Content

Total carbohydrate content was determined by difference. On a dry weight basis, the amounts of ash, proteins, lipids, crude fiber, and moisture were subtracted from the total sample weight [16].

## 2.5. Ash Content Determination

Total ash content was determined using the AOAC method [15], which involved incinerating dried pulp and seed samples at 550°C in an oxidizing atmosphere until a constant residue mass was obtained.

## 2.6. Mineral Content Determination

Mineral contents were determined using an energy-dispersive X-ray fluorescence spectrometer (EDXRF), model EDX-7000 (Shimadzu). Each ash sample was dissolved in 20% nitric acid and then filtered through Whatman filter paper. Calcium and magnesium were quantified by complexometric titration, while total phosphorus was determined colorimetrically at 430 nm using the mineralized extract. Iron was measured using the orthophenanthroline method by colorimetry with a maximum absorption at 510 nm. Copper, sodium, potassium, and manganese contents were determined by atomic absorption spectrophotometry after ash solubilization. All analyses were carried out using standard methods described by AOAC [16].

## 2.7. Preparation of Extracts

Drying of the different samples was carried out in an oven at 45°C for 10 hours, and 100 g of the resulting powder were macerated in 800 mL of distilled water and left to stand for 24 hours at room temperature. After filtration, the extracts were obtained and used for the determination of phytochemical compound and antioxidant activities.

## 2.8. Phytochemical Analysis

### 2.8.1. Total Phenolic Content Determination

Approximately 1 mL of Folin–Ciocalteu reagent was added

to a tube containing 200  $\mu$ L of the aqueous extract of the samples. After vigorous shaking, 800  $\mu$ L of sodium carbonate solution (0.75%) were added, and the tubes were incubated in the dark for 90 minutes. Total phenolic content was expressed as mg gallic acid equivalent (GAE) and calculated from the calibration curve ( $y = 0.0216x$ ) obtained from the linear regression of values derived from the standard solution.

### 2.8.2. Flavonoid Content Determination

A volume of 500  $\mu$ L of the sample was placed in a tube with 1 mL of a 2% aluminium trichloride ( $AlCl_3$ ) solution. After incubation in the dark for 15 minutes at room temperature, absorbance was measured at 430 nm. Flavonoid content was determined using a calibration curve ( $y = 0.0348x - 0.015$ ) obtained from the linear regression of values derived from the standard solution.

### 2.8.3. Tannin Content Determination

A volume of 0.5 mL of extract was added to 1 mL of a 4% vanillin solution in test tubes. After mixing, 750  $\mu$ L of concentrated HCl were added, and the mixture was left to stand at room temperature in the dark for 20 minutes. Absorbance was measured at 500 nm, and tannin content was expressed as mg tannic acid equivalent, based on a calibration curve ( $y = 0.0011x$ ) obtained from the linear regression of values derived from the standard solution.

## 2.9. In Vitro and ex Vivo Antioxidant Activity Evaluation

### 2.9.1. In Vitro Antioxidant Activity Evaluation

#### (i). DPPH Radical Scavenging Assay

In a volume of 800  $\mu$ L of DPPH (2,2-diphenyl-1-picrylhydrazyl) solution, 200  $\mu$ L of sample at different concentrations (2000, 1000, 500, 250, and 125  $\mu$ g/mL) were added, along with vitamin C as a positive control. The mixture was incubated in the dark at room temperature for 30 minutes, and absorbance was measured at 517 nm, protected from light (Rumpf et al., 2023). The obtained absorbance values were used to calculate the radical scavenging activity.

#### (ii). ABTS Assay

ABTS (2,2'-Azinobis (3-Ethylbenzothiazoline 6-Sulfonate)) radical scavenging activity was evaluated according to the method described by Rumpf et al. [17] with slight modifications. The ABTS solution was diluted until an absorbance of 1.5 was obtained at 734 nm. A volume of 5  $\mu$ L of each sample or vitamin C at different concentrations (125, 250, 500, 1000, and 2000  $\mu$ g/mL) was added to 1800  $\mu$ L of ABTS solution. The mixture was incubated in the dark for 15 minutes, and absorbance was measured at 734 nm using a

spectrophotometer against a blank. The percentage of inhibition was then determined.

### (iii). Reducing Power (FRAP: Ferric Reducing Antioxidant Power)

A volume of 1.25 mL of phosphate buffer (0.2 M, pH 6.6) and 1.25 mL of 1% potassium ferricyanide were added to 500  $\mu$ L of the different sample concentrations. After incubation at 50°C for 20 minutes, 1.25 mL of 10% trichloroacetic acid (TCA) solution was added to the reaction mixture. The mixture was then centrifuged at 3000 rpm for 10 minutes. Subsequently, 1.25 mL of distilled water and 250  $\mu$ L of 0.1% FeCl<sub>3</sub> were added to 1.25 mL of the supernatant, and absorbance was measured at 700 nm [17].

### 2.9.2. Ex Vivo Antioxidant Activity Evaluation

It was assessed using sheep erythrocytes according to the method described and modified by Nkouandou et al. [18]. This evaluation provides an approach to simulate physiological conditions and to assess the protective capacity of the extracts against induced oxidative stress.

#### (i). Preparation of Sheep Erythrocytes

Blood from a healthy sheep collected in EDTA tubes was centrifuged at 3600 rpm for 15 minutes, and red blood cells were separated from the plasma. The erythrocytes were washed three times with physiological saline solution (0.9% NaCl). During the final wash, the erythrocytes were centrifuged for 10 minutes to obtain a packed cell preparation, which was then resuspended in physiological saline. Copper sulfate was used to induce hemolysis by promoting the oxidation of membrane lipids and proteins. Then, 500  $\mu$ L of the erythrocyte suspension were mixed with 500  $\mu$ L of physiological solution containing 250  $\mu$ L of sample at different concentrations and 100  $\mu$ L of copper sulfate (10 mM in 0.9% NaCl). The different reaction mixtures were incubated at room temperature for 30 minutes with gentle agitation. After incubation, the reaction mixture was diluted with physiological solution and then centrifuged at 3600 rpm for 5 minutes. Complete hemolysis of the red blood cells was achieved by adding ice-cold distilled water, and the samples were stored at -20°C for further analysis.

#### (ii). Measurement of Oxidative Stress Parameters

##### (a). Malondialdehyde Assay

A volume of 500  $\mu$ L of orthophosphoric acid and 500  $\mu$ L of precipitation mixture (1% thiobarbituric acid in 1% acetic acid) were added to 100  $\mu$ L of homogenate. The reaction mixture

was homogenized and incubated in a water bath at 100°C for 15 minutes, then cooled for 30 minutes. The mixture was centrifuged for 10 minutes, and the absorbance of the supernatant was measured using a spectrophotometer at 532 nm [18].

##### (b). Superoxide Dismutase Assay

Approximately 150  $\mu$ L of homogenates, 500  $\mu$ L of carbonate-bicarbonate buffer (pH 10.2, 0.3 M, pKa 10.3), 250  $\mu$ L of EDTA solution (0.6 mM), and 350  $\mu$ L of distilled water were mixed. The mixture was homogenized, and 250  $\mu$ L of adrenaline (4.5  $\mu$ M) were added to initiate the reaction. Adrenaline auto-oxidation was monitored by measuring optical density at 480 nm after 30 seconds and 180 seconds using a spectrophotometer, following adrenaline addition. SOD activity was expressed as percentage inhibition [18].

##### (c). Catalase Activity Assay

Approximately 375  $\mu$ L of phosphate buffer (pH 7.4) were added to 25  $\mu$ L of tissue homogenate, followed by 100  $\mu$ L of hydrogen peroxide (50 mM). Subsequently, 2 mL of dichromate/acetic acid solution were added to stop the reaction after 60 seconds. All tubes were then incubated at 100°C for 10 minutes [18]. After cooling in an ice bath, optical densities were measured at 570 nm using a spectrophotometer.

### 2.10. Statistical Analyses

Quantitative data were expressed as mean  $\pm$  standard deviation. All chemical analyses were performed in triplicate (n=3). Differences between samples were assessed using Student's *t*-test for pairwise comparisons. Statistical significance was set at P<0.05.

## 3. Results and Discussion

### 3.1. Results

#### 3.1.1. Proximate Composition of Rambutan Pulp and Seed

Table 1 presents the moisture contents, proteins, lipids, carbohydrates, dietary fiber, and ash in the pulp and seed of rambutan per 100 g of fresh matter (FM). It appears that the pulp is characterized by a higher moisture contents compared to the seed. The carbohydrate, lipid, protein, and dietary fiber contents of the seed were significantly higher than those of the pulp, with coverage rates of 12.80%, 28.40%, 17.70%, and 32.80%, respectively. The ash content of the seed was also higher than that of the pulp. The recommended daily intakes were for children aged 1 to 3 years according to FAO/WHO, 2015.

**Table 1.** Proximate Composition of Rambutan Fruit per 100 g of Fresh Matter and Recommended Daily Intakes.

Macronutrients (g/100g FM)	Seed	Pulp	RDI for children (1-3 years) (g/j)
Total carbohydrates	22.95 ± 1.90 <sup>a</sup>	11.59 ± 0.60 <sup>b</sup>	180
Lipids	9.95 ± 0.13 <sup>a</sup>	0.27 ± 0.02 <sup>b</sup>	35
Proteins	2.65 ± 0.09 <sup>a</sup>	0.46 ± 0.02 <sup>b</sup>	15
Crude fibers	4.59 ± 0.07 <sup>a</sup>	3.74 ± 0.16 <sup>b</sup>	14
Ash	10.86 ± 1.13 <sup>a</sup>	2.94 ± 0.20 <sup>b</sup>	/
Moisture contents	49 ± 1 <sup>a</sup>	81 ± 1 <sup>b</sup>	/

These values represent the mean ± standard deviation; n = 3; values in the same row that do not share the same superscript letters are significantly different (P < 0.05); FM = Fresh Matter; RDI = Recommended Daily Intake.

### 3.1.2. Mineral Composition of Rambutan

The mineral contents of the pulp and seed of rambutan are presented in Table 2. The results show that the pulp has a significantly higher phosphorus content (P < 0.05) compared to the seed. In contrast, the seed contains signifi-

cantly higher levels of potassium, calcium, iron, zinc, manganese, and copper than the pulp. These values correspond to contributions to the recommended daily intakes for children aged 1 to 3 years, estimated at 4.60% for potassium, 1.50% for calcium, 8.70% for iron, 2% for zinc, 49.20% for manganese, and 64% for copper, according to FAO/WHO recommendations [19].

**Table 2.** Mineral Content in Rambutan Fruits per 100 g of Fresh Matter.

Minerals (mg/100g)	Seed	Pulp	RDI for children (g/j)
K	37 ± 1 <sup>a</sup>	10 ± 0.10 <sup>b</sup>	800
Ca	10.54 ± 0.18 <sup>a</sup>	5.83 ± 0.55 <sup>b</sup>	700
Fe	0.61 ± 0.001 <sup>a</sup>	0.14 ± 0.11 <sup>b</sup>	7
P	0.91 ± 0.02 <sup>a</sup>	2.15 ± 0.20 <sup>b</sup>	250
Mn	0.59 ± 0.03 <sup>a</sup>	0.10 ± 0.08 <sup>b</sup>	1.20
Zn	0.061 ± 0.008 <sup>a</sup>	0.019 ± 0.007 <sup>b</sup>	3
Cu	0.48 ± 0.01 <sup>a</sup>	0.03 ± 0.003 <sup>b</sup>	0.75

These values represent the mean ± standard deviation; n = 3; values in the same row that do not share the same superscript letters are significantly different (P < 0.05); FM = Fresh Matter; RDI = Recommended Daily Intake.

### 3.1.3. Phytochemical Composition

The phytochemical analysis of the aqueous extracts of rambutan pulp and seed showed that the seed extract contained significantly higher levels of flavonoids and tannins compared

to the pulp (Table 3). The total polyphenol content of the seed was higher than that of the pulp, although the difference was not statistically significant. The presence of polyphenols, flavonoids, and tannins indicates that both parts of rambutan (*Nephelium lappaceum* L.) represent interesting natural sources of bioactive compounds.

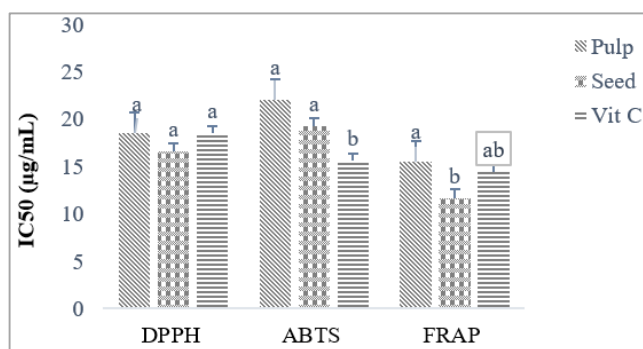
**Table 3.** Phytochemical Compound Contents of *Nephelium lappaceum* L.

Samples	Polyphenols ( $\mu\text{g AGE}/\text{mg}$ )	Flavonoids ( $\mu\text{g QE}/\text{mg}$ )	Tannins ( $\mu\text{g ATE}/\text{mg}$ )
Pulp	449.20 $\pm$ 21.70 <sup>a</sup>	26.60 $\pm$ 4.60 <sup>a</sup>	0.20 $\pm$ 0.04 <sup>a</sup>
Seed	523.04 $\pm$ 61.10 <sup>a</sup>	51.70 $\pm$ 1.35 <sup>b</sup>	0.50 $\pm$ 0.04 <sup>b</sup>

These values are expressed as mean  $\pm$  standard deviation; n = 3. Values within the same row not sharing the same superscript letter are significantly different ( $P < 0.05$ ).

### 3.1.4. In Vitro Antioxidant Activity of Rambutan Extracts

The results presented in Figure 1 showed that both pulp and seed extracts of rambutan exhibited antioxidant activity, as assessed by the DPPH method ( $\text{IC}_{50}$  values of 18.6 and 16.6  $\mu\text{g}/\text{mL}$  for pulp and seed, respectively) and the ABTS assay ( $\text{IC}_{50}$  values of 22.1 and 19.2  $\mu\text{g}/\text{mL}$  for pulp and seed, respectively), although these  $\text{IC}_{50}$  values were slightly lower than that of vitamin C used as a reference. For the FRAP assay,  $\text{IC}_{50}$  values were 15.5  $\mu\text{g}/\text{mL}$ , 11.7  $\mu\text{g}/\text{mL}$ , and 14.5  $\mu\text{g}/\text{mL}$  for pulp, seed, and vitamin C, respectively. The seed exhibited the highest reducing activity.

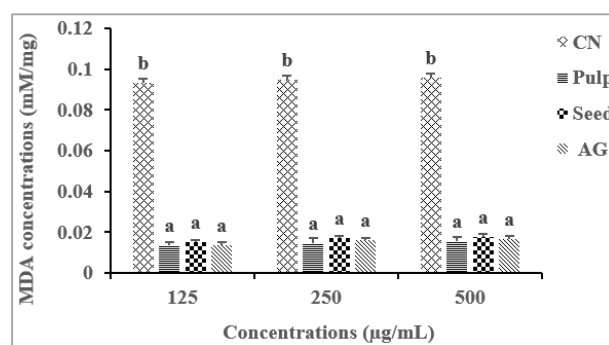


**Figure 1.** Inhibitory Concentration of Different Extracts in Antioxidant Assays.

### 3.1.5. Ex Vivo Antioxidant Activity of Rambutan Extracts

#### (i). Malondialdehyde Concentration

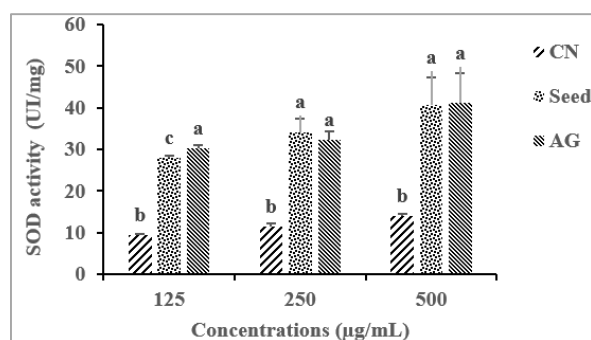
Figure 2 illustrates the effect of rambutan pulp and seed extracts on malondialdehyde (MDA) levels. The highest MDA values were recorded in the negative control (CN), with an average concentration of approximately 0.09 mM/mg for the three tested concentrations, indicating strong lipid oxidative degradation.



**Figure 2.** Effect of Rambutan Pulp and Seed extracts on MDA Concentration.

#### (ii). Superoxide Dismutase Activity

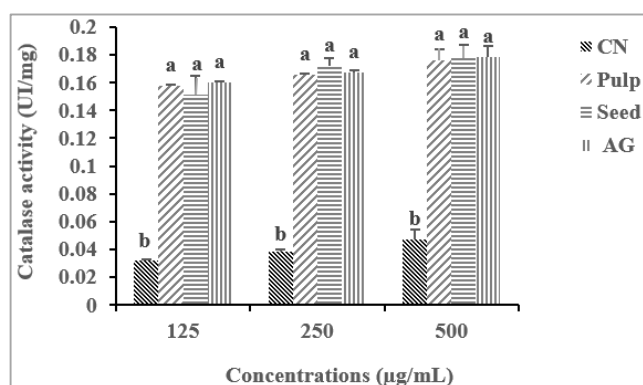
Figure 3 shows the evolution of superoxide dismutase (SOD) activity as a function of the concentrations of rambutan pulp and seed extracts. An increase in enzymatic activity was observed with increasing extract concentrations (125, 250, and 500  $\mu\text{g}/\text{mL}$ ), indicating a dose-dependent effect. The negative control (CN) showed low and constant values, reflecting an unbalanced oxidative stress state. At 125  $\mu\text{g}/\text{mL}$ , both pulp and seed extracts exhibited similar activities (26.7 and 27.93 U/mg, respectively), which were significantly higher than that of the negative control (9.31 U/mg), but slightly lower than that of the standard gallic acid (30.4 U/mg). At 250 and 500  $\mu\text{g}/\text{mL}$ , SOD activity increased further, reaching approximately 38 to 41.3 U/mg in the treated groups.



**Figure 3.** Effect of Rambutan Pulp and Seed Extracts on Superoxide Dismutase Activity.

### (iii). Catalase Activity

The results of catalase activity are presented in Figure 4. The enzyme activity was very low in the negative control (CN), with values close to 0.04 U/mg. In contrast, the rambutan pulp and seed extracts, as well as gallic acid (AG), showed significantly higher activities, with values of 0.15, 0.16, and 0.17 U/mg at concentrations of 125, 250, and 500  $\mu\text{g/mL}$ , respectively.



**Figure 4.** Effect of rambutan pulp and seed extracts on catalase activity.

## 3.2. Discussion

The proximate composition of rambutan pulp presented in Table 1 showed a high moisture content, which is consistent with the values reported by Mahmood et al. [11], ranging from 78.5 to 84.7 g/100 g fresh matter (FM), but higher than those reported by Afzaal et al. [8], which varied between 34.25 and 34.6 g/100 g FM. These differences may be attributed to varietal differences or the ripening stage of the rambutan fruit [20, 21]. Its carbohydrate content was  $11.59 \pm 0.2$  g/100 g FM, which is lower than the values reported by Afzaal et al. [8], ranging from 33.63 to 61.5 g/100 g FM, and also lower compared to some commonly consumed fruits such as mango, which contains higher carbohydrate levels (14.98 g/100 g) [22]. This carbohydrate content could contribute approximately 6.4% of the recommended dietary allowance (RDA) for children aged 1 to 3 years, according to FAO/WHO [19]. The dietary fiber content was higher than the 1.6 g/100 g reported in mango [22] and the  $3.10 \pm 0.06$  g/100 g found in papaya [23]. This high fiber content could meet approximately 26.70% of the recommended daily intake (RDI), giving the pulp a nutritional interest for digestive regulation and the prevention of metabolic and gastrointestinal disorders [24]. The lipid and protein contents (Table 1) were low compared to the ranges of 18.20 to 36.10 g/100 g FM and 2.21 to 2.91 g/100 g FM reported by Biswas (2021), respectively, indicating the low energy value of the pulp. The ash content ( $2.94 \pm 0.15$  g/100 g FM) was lower than the values reported by Afzaal et al. [8], which ranged from 11.05 to 14.2 g/100 g FM, suggesting a lower mineral density. The proximate composition of

rambutan seed (Table 1) showed a carbohydrate content of  $22.9 \pm 1.53$  g/100 g fresh matter (FM), corresponding to 12.7% of the recommended daily intake for children aged 1 to 3 years. This value falls within the range of 15.5 to 18.99 g/100 g FM reported by Afzaal et al. [8]. These variations are likely due to differences in extraction methods and growing conditions. Its lipid content was higher than the values reported by Mahmood et al. [11], which ranged from 0.45 to 1.03 g/100 g FM, but lower than the  $38.90 \pm 0.32$  g/100 g FM reported by Rakariyatham et al. [25], confirming the role of the seed as an important energy reserve for embryonic development and suggesting that it could contribute approximately 16.7% of the recommended daily intake. The protein content was lower than the  $12.4 \pm 0.22$  g/100 g FM reported by Rakariyatham et al. [25], highlighting the potential of the seed as a complementary source of plant proteins. The fiber content of the seed was higher compared to the values reported by Afzaal et al. [8] and Biswas [20], which ranged from 0.29 to 2.70 g/100 g FM. These high fiber levels could cover nearly 32.8% of the recommended daily intake (RDI), thereby contributing effectively to intestinal transit regulation and the prevention of digestive disorders [24]. The ash content was higher than the  $2.26 \pm 0.42$  g/100 g fresh matter reported by Rakariyatham et al. [25], suggesting a notable mineral density. Compared with other commonly consumed fruits such as mango and papaya [22, 23], rambutan is characterized by a low lipid, carbohydrate, and protein content, but a high fiber content. This makes it suitable for dietary regimens aimed at low energy density and supports intestinal transit regulation and proper digestive function [24].

Table 2 shows that the pulp has a significantly higher phosphorus content ( $p < 0.05$ ) compared to the seed. In contrast, the seed contains significantly higher levels of potassium, calcium, iron, zinc, manganese, and copper than the pulp. These levels highlight the potential nutritional role of the seed, as potassium contributes to the regulation of blood pressure and heart rhythm, while calcium helps prevent osteoporosis and maintains bone density, particularly in elderly individuals and postmenopausal women [26]. The iron content could cover approximately 8.7% of the recommended daily intake for children aged 1 to 3 years, potentially supporting hemoglobin formation and playing a crucial role in oxygen transport in the blood as well as energy metabolism [27]. Meanwhile, the zinc and copper contents in the seed may contribute to nucleic acid synthesis and play an essential role in immune function and wound healing [28]. The potassium, calcium, and manganese contents to the seed was also lower than those reported by Torgbo et al. [29], with values of  $525.10 \pm 7.87$  mg/100 g,  $452.89 \pm 3.62$  mg/100 g, and  $17.69 \pm 0.53$  mg/100 g, respectively. These differences may be attributed to varietal differences or the ripening stage of the rambutan fruit. The mineral content in rambutan fruits (Table 2) indicate that the seed has a higher mineral density than the pulp, highlighting its nutritional interest in human diets, particularly in the context of dietary strategies aimed at improving micronutrient intake.

The antioxidant activity of rambutan pulp and seed extracts was evaluated using three complementary methods (DPPH, ABTS, and FRAP). The IC<sub>50</sub> values for pulp, seed, and vitamin C in the DPPH assay were not significantly different (Figure 1). These results indicate a good free radical scavenging capacity against DPPH, with a slight trend toward higher efficiency in the seed extract. For the ABTS assay, the IC<sub>50</sub> values obtained for pulp, seed, and vitamin C were also not significantly different. Although these differences were not statistically significant, the seed extract showed a more pronounced antioxidant activity, close to that of vitamin C, likely due to the presence of water-soluble flavonoids such as quercetin and catechins, which are known for their effective interaction with the ABTS<sup>+</sup> radical [3]. The performance of vitamin C in this assay may be explained by a favorable interaction kinetic with the ABTS<sup>+</sup> radical. The FRAP assay highlighted a lower IC<sub>50</sub> value for the seed compared to the pulp and vitamin C, indicating a stronger reducing power of the seed. This result could be attributed to a better synergy between hydrophilic and lipophilic compounds in the seed, facilitating electron transfer to ferric iron, as well as its richness in phenolic acids such as gallic acid [3].

Table 3 shows that the seed contained significantly higher levels of flavonoids and tannins compared to the pulp. This richness in phenolic compounds in the seed may be related to the presence of protective structures such as the cuticle and sclereid cells, which play a role in chemical defense against biotic and abiotic stresses [11]. Total polyphenols, although slightly more concentrated in the seed than in the pulp, did not show a statistically significant difference. This may be explained by the diversity of phenolic compounds as well as the influence of ripening stage, geographical origin, and extraction methods [3, 25]. The high content of flavonoids and tannins in the seed appears to be directly related to its stronger antioxidant capacity in radical scavenging assays (DPPH and ABTS) and in reducing power (FRAP), as these compounds are well known for their effectiveness in neutralizing reactive oxygen species [30]. The pulp, on the other hand, also exhibited notable antioxidant activity, attributable to its soluble polyphenols, particularly gallic acid and its derivatives, as well as anthocyanins [8, 25].

The *ex vivo* antioxidant activity highlighted the potential protective effect of the tested rambutan pulp and seed extracts. Exposure of erythrocytes to copper resulted in a significant increase in malondialdehyde (MDA) levels in the negative control group (Figure 2), reflecting substantial lipid peroxidation. In contrast, co-incubation with the tested extracts significantly reduced MDA levels, indicating a marked inhibition of lipid peroxidation. At 500 µg/mL, MDA values were very close to those observed in the reference group, suggesting an effect comparable to that of a standard antioxidant such as catechin used in the comparative study by Pereira-Freire et al. [30]. In parallel, superoxide dismutase (SOD) activity increased in a dose-dependent manner with the extracts, reaching approximately 41 U/mg at 500 µg/mL, a level higher than

that observed at 125 µg/mL and slightly higher than that of the reference compound (Figure 3). This enzymatic induction is consistent with the findings reported by Nkouandou et al. [18], where SOD activity also increased in the presence of polyphenolic extracts in a concentration-dependent manner. In contrast to SOD, catalase activity did not show significant variation across the different extract concentrations, although the measured values were markedly higher than those of the negative control. This activity plateau observed from 125 µg/mL (≈ 0.15 U/mg) may indicate maximal enzyme activation at the lowest tested concentration, or alternatively suggest saturation of the enzymatic system beyond a certain threshold.

## 4. Conclusion

At the end of this study, it appears that both rambutan pulp and seed possess nutritional and bioactive properties; however, the seed exhibits a higher nutritional density than the pulp. The obtained results position this fruit as a potential source for the development of dietary supplements, functional food ingredients, and therapeutic formulations, as well as for dietary strategies aimed at improving micronutrient intake.

## Abbreviations

ABTS	2,2'-Azinobis (3-Ethylbenzothiazoline 6-Sulfonique Acide)
AlCl <sub>3</sub>	Aluminium Trichloride
DPPH	2,2-diphenyl-1-picrylhydrazyl
FM	Fresh Matter
FRAP	Ferric Reducing Antioxidant Power
GAE	Gallic Acid Equivalent
MDA	Malondialdehyde
RDI	Recommended Daily Intake
TCA	Trichloroacetic Acid

## Acknowledgments

The authors are grateful to the people who facilitated this study. They also express their sincere thanks to all the members of the Biochemistry Department laboratory for their cooperation and support.

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## Conflicts of Interest

The authors declared no conflicts of interest.

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